

# Chemistry

## Recommended Time Table

### Class-10

Chapter Names	Concepts
<b>1. Atomic Structure (4 hours)</b>	<ul style="list-style-type: none"> <li>• Rutherford's, Bohr's • Sommerfeld's models <span style="float: right;">(1 hour)</span></li> <li>• Quantum numbers • EC (of atoms) de Broglie's theory</li> </ul>
	<ul style="list-style-type: none"> <li>Heisenberg's principle <span style="float: right;">(1 hour)</span></li> <li>• Problems and exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>2. Periodic Table (4 hours)</b>	<ul style="list-style-type: none"> <li>• Mendeleev &amp; Modern periodic table • Classification of elements <span style="float: right;">(1 hour)</span></li> <li>• Gradation in periodic properties <span style="float: right;">(1 hour)</span></li> <li>• Exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>3. Chemical Bonding (4 hours)</b>	<ul style="list-style-type: none"> <li>• Bonds - Ionic • Covalent • Coordinate covalent <span style="float: right;">(1 hour)</span></li> <li>• Overlapping of orbitals • shapes of molecules • Metallic bond <span style="float: right;">(1 hour)</span></li> <li>• Exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubt Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>4. Mole Concept and Stoichiometry (5 hours)</b>	<ul style="list-style-type: none"> <li>• At.wt • Mol.wt. • Mole <span style="float: right;">(1 hour)</span></li> <li>• Stoichiometry <span style="float: right;">(1 hour)</span></li> <li>• Problems and exercises <span style="float: right;">(2 hours)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>5. Electrochemistry (3 hours)</b>	<ul style="list-style-type: none"> <li>• Electrolysis • Faraday's laws • Electrochemical cells <span style="float: right;">(1 hour)</span></li> <li>• Problems and exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>6. Chemical Kinetics and Chemical Equilibrium (4 hours)</b>	<ul style="list-style-type: none"> <li>• Rate of reaction • Rate laws • Collision theory</li> <li>• Factors affecting rate <span style="float: right;">(1 hour)</span></li> <li>• Chemical equilibrium • Factors affecting it <span style="float: right;">(1 hour)</span></li> <li>• Problems and exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>7. Acids, Bases and Salts (4 hours)</b>	<ul style="list-style-type: none"> <li>• Preparation • Properties of acids • bases &amp; salts • Arrhenius theory • <math>K_w</math>, pH <span style="float: right;">(1 hour)</span></li> <li>• Problems and exercises <span style="float: right;">(2 hours)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>8. Compounds of Nitrogen (3 hours)</b>	<ul style="list-style-type: none"> <li>• Ammonia Preparation • Properties • Uses</li> <li>• Oxides of nitrogen • <math>\text{HNO}_3</math> <span style="float: right;">(1 hour)</span></li> <li>• Exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>
<b>9. Compounds of Sulphur (3 hours)</b>	<ul style="list-style-type: none"> <li>• Preparation • Properties and Uses, <math>\text{H}_2\text{S}</math>, <math>\text{SO}_2</math>, <math>\text{SO}_3</math>, <math>\text{H}_2\text{SO}_3</math>, <math>\text{H}_2\text{SO}_4</math> <span style="float: right;">(1 hour)</span></li> <li>• Exercises <span style="float: right;">(1 hour)</span></li> <li>• Doubts Clarification <span style="float: right;">(1 hour)</span></li> </ul>

Chapter Names	Concepts
<b>10. Compounds of Chlorine</b> (3 hours)	<ul style="list-style-type: none"> <li>• Preparation • Properties and Uses of NaCl, HCl (1 hour)</li> <li>• Bleaching powder (1 hour)</li> <li>• Exercises (1 hour)</li> <li>• Doubts Clarification (1 hour)</li> </ul>
<b>11. Organic Chemistry-I</b> (4 hours)	<ul style="list-style-type: none"> <li>• Unique features of carbon • Nomenclature • Hydrocarbons (2 hours)</li> <li>• Preparation • Properties • Uses (1 hour)</li> <li>• Exercises (1 hour)</li> <li>• Doubts Clarification (1 hour)</li> </ul>
<b>12. Organic Chemistry-II</b> (5 hours)	<ul style="list-style-type: none"> <li>• Nomenclature of various organic compounds • their preparation (1 hour)</li> <li>• Properties and Uses (1 hour)</li> <li>• Esters, Ether, Biomolecules (2 hours)</li> <li>• Exercises (1 hour)</li> <li>• Doubts Clarification (1 hour)</li> </ul>
<b>13. Metallurgy</b> (4 hours)	<ul style="list-style-type: none"> <li>• Metallurgical processes (1 hour)</li> <li>• Extraction of Al, Zn, Fe (1 hour)</li> <li>• Exercises (1 hour)</li> <li>• Doubts Clarification (1 hour)</li> </ul>

### Chemistry

Term - 1 (MCQ's: 50)

Chemistry: Chapters. 1 to 6

Term - 2 (MCQ's: 50)

Chemistry: Chapters. 1 to 10

Term - 3 (MCQ's: 50)

Chemistry: Chapters. 1 to 13

### Physics

Term - 1 (MCQ's: 50)

Physics: Chapters. 1 to 4

Term - 2 (MCQ's: 50)

Physics: Chapters. 1 to 6

Term - 3 (MCQ's: 50)

Physics: Chapters. 1 to 8

Chapter Names	Concept	Problems	Total No. of hours
1. Atomic Structure	1	3	4
2. Periodic Table	2	3	4
3. Chemical Bonding	2	3	4
4. Mole Concept & Stoichiometry	2	2	5
5. Electrochemistry	1	3	3
6. Chemical Kinetics & Chemical Equilibrium	1	3	4
7. Acids, Bases & Salts	2	3	4
8. Compounds of Nitrogen	2	2	3
9. Compounds of Sulphur	1	2	3
10. Compounds of Chlorine	2	2	3
11. Organic Chemistry - I	1	2	4
12. Organic Chemistry - II	2	2	5
13. Metallurgy	1	2	4